

**BITTA METALL BILAN MODIFIKATSIYA QILINGAN
KATALIZATORLAR ISHTIROKIDA PROPAN-BUTAN
ARALASHMASINING C-C BOG'INING UZILISHI BILAN BORADIGAN
PARCHALANISHI**

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Tabiiy va neft yo'ldosh gazlarining hamda propan-butan fraktsiyasining asosiy komponentlari C₂-C₄ olefinlarini [1-2], aromatik uglevodorodlar [3-5] va boshqa qimmatbaho neft-kimyo mahsulotlarini [6-11] olish uchun arzon uglevodorod xom ashyosidir.

Mikrog'ovak seolit katalizatorlari quyi C₃-C₄ alkanlarini quyi olefinlar va aromatik uglevodorodlarga aylantirish uchun eng istiqbolli materiallar bo'lishi mumkin, chunki ular turli organik moddalarning dehidrogenatsiyasi, izomerizatsiyasi, degidrotsikllanishi va parchalash reaktsiyalarida yuqori faollik va selektivlikni namoyish etadi [12-15].

Reaksiyaning gaz mahsulotlari xromatografik usulda qo'shimcha termostat bilan jihozlangan issiqlik o'tkazuvchi detektorli "Gazoxrom 3101" xromatografida quyidagi maqbul sharoitda tahlil qilindi: kolonka termostati harorati – 100°C, tashuvchi gaz (havo) oqimining sarfi -35 ml/min, faollashtirilgan ko'mir bilan to'ldirilgan kolonka uzunligi – 1 m, ichki diametri – 3 mm. Miqdoriy tahlil mutlaq darajalash usulida amalga oshirildi.

Rux ta'siri. 1 va 2-jadvallarda propan-butan aralashmasini yuqori haroratlarda parchalash jarayoni uchun yaratilgan katalizator 5Zn-5%CrF₃*5%CoO*NiO*ZrO₂*Na₂SO₄ va 5%CoO*5%NiO*2%ZrO₂*8%Na₂SO₄ katalizatorlar ishtirokida propan-butan aralashmasining yuqori haroratli C-C bog'ining uzilishi bilan boradigan parchalanishi natijalari keltirilgan.

1-jadval

Propan-butan aralashmasini yuqori haroratlarda parchalash jarayoni uchun yaratilgan katalizator 5Zn-5%CoO*5%NiO*2%ZrO₂*8%Na₂SO₄ ishtirokida propan-butan aralashmasining yuqori haroratli C-C bog'ining uzilishi bilan boradigan parchalanishi natijalari

Jarayon sharoiti:

Harorat, °C	600	650	700	750	800
Ta'sirlashuv vaqtি, s	0,24	0,24	0,24	0,24	0,24
Suv bug'i: reagent (dastlabki modda)	0,4:1	0,4:1	0,4:1	0,4:1	0,4:1
Tajriba natijalari: 1 .Unum, % (mass.)					
Gaz, shu jumladan.					
H ₂	99,43	99,21	98,65	98,05	95,93
CH ₄	0,06	0,21	0,66	1,73	1,94
C ₂ H ₄	2,07	4,30	9,02	18,26	26,89
C ₃ H ₆	1,67	7,27	14,47	25,44	31,20
ΣC ₄ H ₈	0,33	1,94	7,89	14,48	13,01
smola	0,06	0,32	0,39	0,68	1,94
Qurum moddalar	0,48	0,68	1,04	1,50	3,37
2.Unum	0,09	0,121	0,31	0,45	0,70
Σ To'yinmagan uglevodorodlari C ₂ -C ₄ % (mass.)	2,07	9,66	22,98	41,19	47,37
3.Konversiya darajasi, % (mass.)	3,74	14,08	33,39	62,89	80,31

1-jadvalda keltirilgan natijalar, 5Zn-5%CoO*5%NiO*2%ZrO₂*8%Na₂SO₄ ishtirokida 700-800 °C harorat oralig'ida etilen unumi 5%CoO*5%NiO*2%ZrO₂*8%Na₂SO₄ ishtirokidagi etilen unumi bilan solishtirganda 0,03-0,90% ga yuqori bo'ldi. 5%CoO*5%NiO*2%ZrO₂*8%Na₂SO₄ ni rux bilan transformatsiyasi sinovlarning butun harorat oralig'ida propilen unumining pasayishiga olib keldi. 800 °C da 5Zn-5%CoO*5%NiO*2%ZrO₂*8%Na₂SO₄ da 5%CoO*5%NiO*2%ZrO₂*8%Na₂SO₄ ga nisbatan C₂-C₄ to'yinmagan uglevodorodlar umumiy unumi 2% ga yuqori bo'ldi.

2-jadval

Propan-butan aralashmasini yuqori haroratlarda parchalash jarayoni uchun yaratilgan katalizator 5Zn-5%CrF₃*5%CoO*NiO*ZrO₂*Na₂SO₄ ishtirokida propan-butan aralashmasining yuqori haroratli C-C bog'ining uzilishi bilan boradigan parchalanishi natijalari

Jarayon sharoiti:

Harorat, °C	600	650	700	750	800
Ta'sirlashuv vaqtি, s	0,24	0,24	0,24	0,24	0,24
Suv bug'i: reagent (dastlabki modda)	0,4:1	0,4:1	0,4:1	0,4:1	0,4:1
Tajriba natijalari: 1 .Unum, % (mass.)					
Gaz, shu jumladan.					

H ₂	99,49	99,07	98,75	97,86	96,77
CH ₄	0,04	0,2	0,34	0,97	1,89
C ₂ H ₄	1,64	1,85	3,92	12,51	23,92
C ₃ H ₆	0,5	1,96	7,05	21,2	35,84
ΣC ₄ H ₈	0,21	0,38	3,8	13,02	15,67
smola	izlari	0,01	0,125	0,81	1,92
Qurum moddalar	0,4	0,79	0,9	1,63	2,5
2.Unum	0,121	0,124	0,35	0,51	0,73
Σ To‘yinmagan uglevodorodlari C ₂ -C ₄	0,71	2,39	11,14	35,45	54,79
% (mass.)	1,85	4,38	15,84	50,70	83,98
3.Konversiya darajasi, %(mass.)					

2-jadvalda eksperimental tadqiqotlar natijasiga asoslanib keltirilgan ma'lumotlarga ko'ra, 5%CrF₃*5%CoO*NiO*ZrO₂*Na₂SO₄ ga nisbatan 5Zn-5%CrF₃*5%CoO*NiO*ZrO₂*Na₂SO₄ ning eng yuqori katalitik faolligi 700÷750 °C haroratda kuzatilgan. Shunday qilib, etilen, propilen va C₂-C₄ to‘yinmagan uglevodorodlar umumi unumining 4,33 - 4,9% , 1,94 - 2,68% (mas.) va 7,34-7,39% mos ravishda oshdi. 600-650 °C va 800 °C haroratida transformatsiyalanmagan 5%CrF₃*5%CoO*NiO*ZrO₂*Na₂SO₄ ishtirokida maqsadli mahsulotlarning unumi yuqori bo'ldi. 700÷750 °C harorat oralig'ida 5Zn-5%CrF₃*5%CoO*NiO*ZrO₂*Na₂SO₄ ishtirokida 5%CrF₃*5%CoO*NiO*ZrO₂*Na₂SO₄ natijalari bilan taqqoslaganda, maqsadli mahsulotlar unumining oshishi bilan bir qatorda, konversiya darajasining 10,07 - 14,0% ga oshganligini ta'kidlash kerak. Reaksiya yonaki mahsulotlarining unumi biroz oshdi.

ADABIYOTLAR

- Choudhary, V.R.; Mantri, K.; Sivadarayana, C. Influence of zeolite factors affecting zeolite acidity on the propane aromatization activity and selectivity of Ga/H-ZSM-5. *Microporous Mesoporous Mater.* 2000, 37, 1–8. [CrossRef]
- Erofeev, V.I.; Trofimova, A.S.; Koval, L.M.; Ryabov, Y.V. Acidity and catalytic properties of Cu-ZSM-5 in conversion of lower alkanes. *Russ. J. Appl. Chem.* 2000, 73, 2057–2061.
- Trofimova, A.S.; Koval, L.M.; Erofeev, V.I. Synthesis of Lower Olefins from C₃–C₄ Alkanes on ZSM-5 Цеолитес Modified with Alkali Metals. *Russ. J. Phys. Chem.* 2000, 74, S537–S540.

4. Xiao, H.; Zhang, J.; Wang, P.; Zhang, Z.; Zhang, Q.; Xie, H.; Yang, G.; Han, Y.; Tan, Y. Mechanistic insight to acidity effects of Ga/HZSM-5 on its activity for propane aromatization. *RSC Adv.* 2015, 112, 92222–92233. [CrossRef]
5. Choudhary, C.V.R.; Panjala, D.; Banerjee, S. Aromatization of propene and n-butene over H-galloaluminosilicate (ZSM-5 type) цеолите. *Appl. Catal. A Gen.* 2002, 231, 243–251. [CrossRef]
6. Trofimova, A.S.; Erofeev, V.I.; Koval, L.M. The Preparation of the lower olefins from C₃–C₄ Alkanes on ZSM-5 Цеолитes modified by Lithium. *Russ. J. Phys. Chem.* 2002, 76, 922–925.
7. Erofeev, V.I.; Adyaeva, L.V. Transformations of straight-run naphthas on indium-modified pentasil. *Russ. J. Appl. Chem.* 2003, 76, 1083–1088. [CrossRef]
8. Guo, J.; Lou, H.; Zhao, H.; Zheng, L.; Zheng, X. Degydrogenation and aromatization of propane over rhenium-modified HZSM-5 catalyst. *J. Mol. Catal. A Chem.* 2005, 239, 222–227. [CrossRef]
9. Caeiro, G.; Carvalho, R.H.; Wang, X.; Wang, X.; Lemos, M.A.N.D.A.; Lemos, F.; Guisnet, M.; Ribeiro, F.R. Activation of C₂–C₄ alkanes over acid bifunctional цеолите catalysts. *J. Mol. Catal. A Chem.* 2006, 255, 131–158. [CrossRef]
10. Chang, F.; Wei, Y.; Liu, X.; Zxao, Y.; Xu, L.; Sun, Y.; Zhang, D.; He, Y.; Liu, Z. A mechanistic investigation of the coupled reaction of n-hexane and methanol over HZSM-5. *Appl. Catal. A Gen.* 2007, 328, 163–173. [CrossRef]
11. Asachenko, E.V.; Rodina, O.V.; Ordomskii, V.V.; Gurev, Y.V.; Ivanova, I.I. Specifics of the deactivation of acid and zinc-containing propane aromatization catalysts. *Petroleum Chem.* 2008, 48, 100–104. [CrossRef]
12. Vosmerikova, L.N.; Barbashin, Y.E.; Vosmerikov, A.V. Catalytic aromatization of ethane on zinc-modified цеолитes of various framework types. *Pet. Chem.* 2014, 54, 420–425. [CrossRef]
13. Fayzullayev, N. I.; Umirzakov, R. R. To obtain acetone by spontaneously hydration of acetylene. *ACS National Meeting Book of Abstracts.* 2005. Vol. 229, pp. U598-U598. Web of Science Core Collection <https://www.webofscience.com/wos/woscc/full-record/WOS:000235066602537>.
14. Muradov, K. M., Fayzullayev, N. I., & Zohidov, K. A. Investigation of influence of various factors to oxidative condensation of methane in C₂-hydrocarbons. In *Abstracts of Papers of the American Chemical Society.* 2003. Vol. 226, pp. U258-U259. 1155 16TH ST, NW, Washington, DC 20036 USA: Amer Chemical Soc. <https://www.webofscience.com/wos/woscc/full-record/WOS:000187062501250>.
15. Fayzullaev, N. Gas chromatographic study of catalytic steam-phase hydration of acetylene. In *Abstracts of Papers of the American Chemical Society.* 2003. Vol. 225, pp. U112-U112. 1155 16TH ST, NW, Washington, DC 20036 USA: Amer Chemical Soc. <https://www.webofscience.com/wos/woscc/full-record/WOS:000187917800439>.